

# Manufacturing Of Cement

PORTLAND CEMENT

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### Points To Discuss...

- What is CEMENT?
  - Introduction
  - Diff. B/w Cement and Portland Cement
  - Components Of Portland Cement
- History of PORTLAND CEMENT.
- Manufacturing of PORTLAND CEMENT.
  - Components
  - Processes
    - Dry Process
    - Wet Process

### INTRODUCTION TO CEMENT

- *Cement* is a fine mineral powder manufactured with very precise processes. Mixed with water, this powder transforms into a paste that binds and hardens when submerged in water. Because the composition and fineness of the powder may vary, cement has different properties depending upon its makeup.
- *Cement* is mainly used as a binder in concrete, which is a basic material for all types of construction, including housing, roads, schools, hospitals, dams and ports, as well as for decorative applications (for patios, floors, staircases, driveways, pool decks) and items like tables, sculptures or bookcases

### DIFFERENCE BETWEEN..

#### CEMENT

• A cement is a binder, a substance used in construction that sets, hardens and adheres to other materials, binding them together. Cement is seldom used solely, but is used to bind sand and gravel (aggregate) together. Cement is used with fine aggregate to produce mortar for masonry, or with sand and gravel aggregates to produce concrete.

### **PORTLAND CEMENT**

- Portland cement is the most common type of cement in general used around the world as a basic ingredient of concrete and mortar.
- It is a fine <u>powder</u>, produced by heating limestone and clay minerals in a <u>kiln</u> to form <u>clinker</u>, <u>grinding</u> the clinker, and adding small amounts of other materials.

## Components of **PORTLAND CEMENT**

### Basic Chemical Components

- Calcium (Ca)
- Silicon (Si)
- Aluminum (Al)
- Iron (Fe)

### Typical Raw Materials

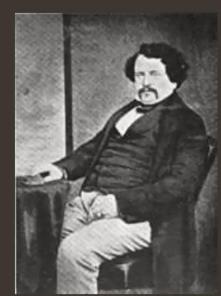
- Limestone (CaCO<sub>3</sub>)
- Sand (SiO<sub>2</sub>)
- Shale, Clay (SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>)
- Iron Ore/Mill Scale (Fe<sub>2</sub>O<sub>3</sub>)

### HISTORY of Portland Cement

Portland cement was developed from natural cements made in Britain beginning in the middle of the 18th century.

Joseph Aspdin took out a patent in 1824 for "Portland Cement," a material he produced by firing finely-ground clay and limestone until the limestone was calcined.

He called it **PORTLAND CEMENT** because the concrete made from it looked like **Portland** stone, a widely-used building stone in England.



Joseph Aspdin

Portland cement is made by heating raw materials rich in oxides of silicon, calcium, aluminum and iron to temperatures of around 1200 - 1400o C.

# MANUFACTURING Of PORTLAND CEMENT

The chemical reactions that occur within the partially molten mass result from the formation of the four main cement materials.

•	Tricalcium silicate	$(CaO)_3 \cdot SiO_2$	$C_3S$	45-75%
•	Dicalcium silicate	$(CaO)_2 \cdot SiO_2$	$\mathrm{C}_2\mathrm{S}$	7-32%
•	Tricalcium aluminate	$(CaO)_3 \cdot Al_2O_3$	$C_3A$	0-13%
•	Tetracalcium alumino-ferrite	$(CaO)_4 \cdot Al_2O_3 \cdot Fe_2O_3$	$\mathrm{C_4AF}$	0-18%
	Gypsum	CaSO. • 2H <sub>2</sub> O	CSH.	2-10%

## Manufacturing Process Of Portland Cement

### > Step 1. Mixing Of Raw Materials

The major raw materials used in the manufacture of cement are Calcium, Silicon, Iron and Aluminum. These minerals are used in different form as per the availability of the minerals.

The <u>Calcareous</u> and <u>Argillaceous</u> Materials used in Manufacturing Process of Portland Cement are shown in the Table in the Next Slide.

### Table shows the raw materials for Portland cement manufacture.

Calcareous Materials	Argillaceous Materials				
Calcium	Silicon	Aluminum	Iron		
Limestone	Clay	Clay	Clay		
Marl	Marl	Shale	Iron ore		
Calcite	Sand	Fly ash	Mill scale		
Aragonite	Shale	Aluminum ore refuse	Shale		
Shale	Fly ash		Blast furnace dust		
Sea Shells	Rice hull ash				
Cement kiln dust	Slag		1		

## Manufacturing Processes Of Portland Cement

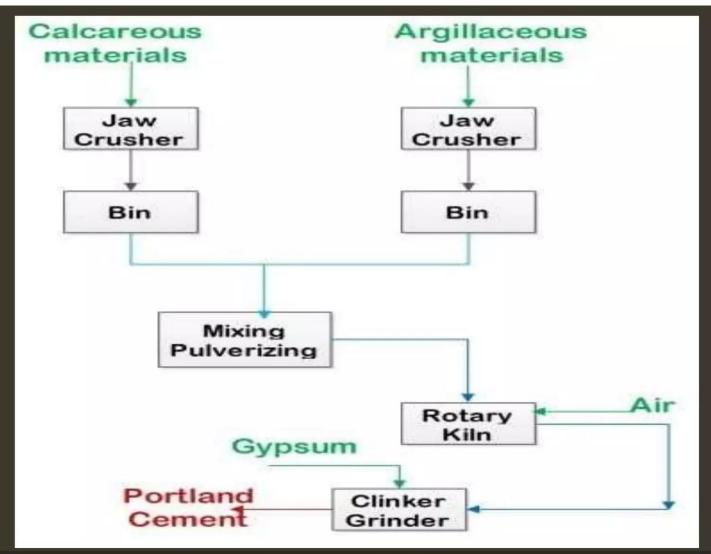
There are **2-Types** of Processes that can be used to Manufacture **Portland Cement.** 

Which are:

- □ Wet Process
- □ Dry Process

### DRY Process

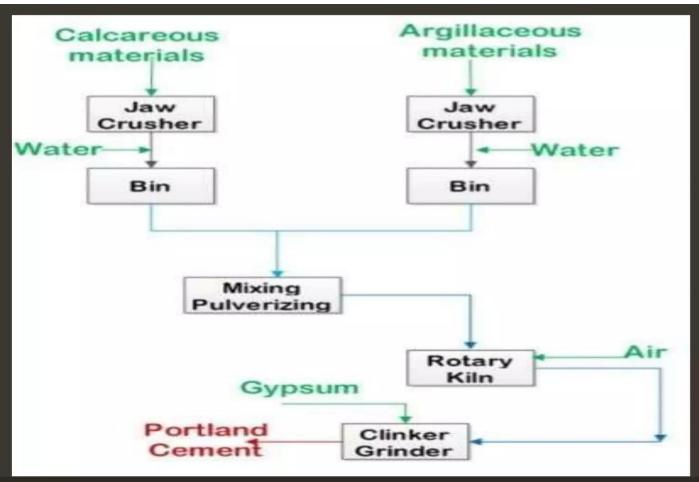
- The both calcareous and argillaceous raw materials are firstly crushed in the gyratory crushers to get 2-5cm size pieces separately. The crushed materials are again grinded to get fine particles into ball or tube mill.
- Each finely grinded material is stored in hopper after screening. Now these powdered minerals are mixed in required proportion to get dry raw mix which is then stored in silos and kept ready to be sent into rotary kiln. Now the raw materials are mixed in specific proportions so that the average composition of the final product is maintained properly.



Manufacturing of cement By DRY process

### WET Process

- The raw materials are firstly crushed and made into powdered form and stored in silos. The clay is then washed in washing mills to remove adhering organic matters found in clay.
- The powdered limestone and water washed clay are sent to flow in the channels and transfer to grinding mills where they are completely mixed and the paste is formed, i.e., known as slurry.
- The grinding process can be done in ball or tube mill or even both. Then the slurry is led into collecting basin where composition can be adjusted. The slurry contains around 38-40% water that is stored in storage tanks and kept ready for the rotary kiln.



Manufacturing of cement By WET process

Comparison of dry process and wet process of Cement Manufacture.

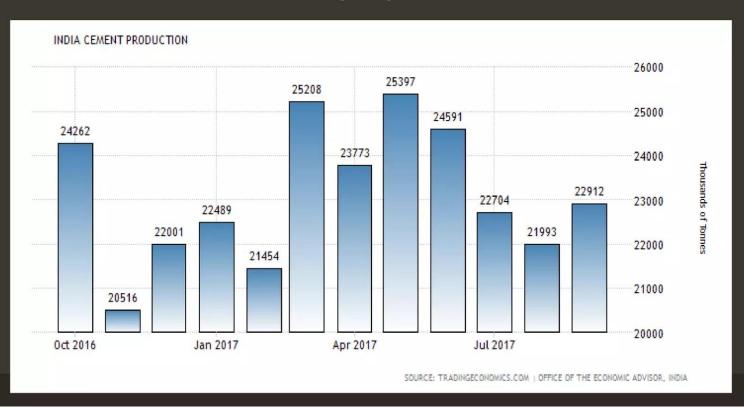
Criteria	Dry Process	Wet Process
Hardness of raw material	Quite hard	Any type of raw material
Fuel consumption	Low	High
Time of process	Lesser	Higher
Quality	Inferior quality	Superior quality
Cost of production	High	Low
Overall cost	Costly	Cheaper
Physical state	Raw mix (solid)	Slurry (liquid)

### > Step 2. Grinding Of Clinkers

- The cooled clinkers are received from the cooling pans and sent into mills. The clinkers are grinded finely into powder in ball mill or tube mill. Powdered gypsum is added around 2-3% as retarding agent during final grinding. The final obtained product is cement that does not settle quickly when comes in contact with water.
- After the initial setting time of the cement, the cement becomes stiff and the *gypsum* retards the dissolution of *tri-calcium aluminates* by forming *tricalcium sulfoaluminate* which is insoluble and prevents too early further reactions of setting and hardening.

 $3\text{CaO.Al}_2\text{O}_3 + \text{xCaSO}_4.7\text{H}_2\text{O} = 3\text{CaO.Al}_2\text{O}_3.\text{xCaSO}_4.7\text{H}_2\text{O}$ 

- > Step 3. Storage and Packaging
  - The grinded cement is stored in silos, from which it is marketed either in container load or 50kg bags.



# Manufacturing Process Of CEMENT

